SYNTHESIS OF ENANTIOMERICALLY PURE 2,3-DISUBSTITUTED ISOXAZOLIDIN-5-ONES

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Abstract. Conjugate addition-rearrangement of N-substituted hydroxylamines to α,β -unsaturated sugar δ -lactones provides a short and effective route to title compounds.

Despite the interesting heterocyclic skeleton of isoxazolidin-5-ones, having a potential synthetic and pharmacological value, there are only a few reports on their synthesis.^{1,2,3} In recent years conjugate addition of *N*-substituted hydroxylamines to α,β -unsaturated esters, followed by cyclization of the adduct to isoxazolidin-5-ones, has attracted attention of several laboratories.^{2,3,4}

In the course of our studies leading to β -lactams from α,β -unsaturated sugar lactones, we have found that benzylhydroxylamine can be added to the double bond of the lactone with high stereoselectivity.⁵ On the other hand, in the presence of formaldehyde, hydroxylamine has been shown to react with <u>1</u> to stereospecifically produce bicyclic compounds <u>2</u>, most likely via the Michael adduct <u>3</u>.⁶ These results have prompted us to investigate the synthesis of 2,3-disubstituted isoxazolidin-5-ones <u>4</u>, based on addition of hydroxylamines <u>5</u> to compounds <u>1</u>.



The lactone <u>1</u> (0.5 mmol) prepared by the known method⁷ was dissolved in abs. ethanol (5 ml) and was treated with <u>5</u> (0.5 mmol) at r.t. for 3 hrs. After evaporation of solvent, the adduct was isolated by chromatography or crystallization in a good or moderate yield; subsequently it was acetylated or silylated by a standard method and then identified spectrally and analytically.⁸

Addition of hydroxylamine 5 proceeds stereospecifically anti with respect to the terminal acetoxymethyl group (axial approach⁹), producing reactive adducts 3 which have a N-hydroxyl group suitably located for reacting with the lactone carbonyl group. The geometry of 3 plays a decisive role in easy formation of the isoxazolidin-5-one ring. Opening of the lactone ring produces compounds 4 with S-configuration at the C-3 isoxazolidine carbon atom when R^2 and R^3 are hydrogen atoms, whereas the R-configuration is obtained when R^2 or R^3 is OAc group. Determination of the configuration of 4 was based on our earlier assignments made for the O-benzylhydroxylamine adduct 6^5 and on X-ray crystal analysis of compound 2 ($R^3=H$)⁶.

The procedure described herein presents a general route for the preparation of enantiomerically pure 2,3-disubstituted isoxazolidin-5-ones which can be utilized in the synthesis of selected structures, e.g. B-amino acids and B-lactams. It is also noteworthy, that this route does not require a strong base catalyzed rearrangement of the hydroxylamine Michael adduct.2,3,4

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- 8. Selected analytical and spectral data for representative compounds 4: Selected analytical and spectral data for representative compounds 4: $\frac{4}{4}$ (R¹=R²=R³=H, R⁴=TBDPS, R⁵=CH₃): 78%; m.p. 70-72°C; IR (film): 1795 cm⁻¹ (C=O); ¹H-NMR (CDCl₃): 1.69, 1.89 (2m, 2H, -CH₂-), 2.57 (dd, 1H, J 10.6, 17.1 Hz, H-4), 2.78 (dd, 1H, J 6.9 Hz, H-4'), 2.85 (s, 3H, NCH₃), 3.31 (bs, 1H, H-3), 3.71, 3.78 (2m, 2H, -CH₂O-). $\frac{4}{4}$ (R¹=CH₂OAc, R²=R³=H, R⁴=TBDMS, R⁵=C₆H₁₁): 73%; syrup; (α)_D -85.0° (c 1.5, CH₂Cl₂); IR (KBr): 1795 (C=O), 1735 cm⁻¹(Ac); ¹H-NMR (CDCl₃): 1.70, 1.88 (2ddd, 2H, -CH₂-), 2.07 (m, 1H, H-3), 3.94 (m, 1H, =CHO-), 3.97-4.05 (m, 2H, -CH₂OAc). $\frac{4}{4}$ (R¹=CH₂OAc, R²=H, R³=OAc, R^{*}=TBDPS, R⁵=CH₂C₆H₄OCH₃): 68%; syrup; (α)_D -51.2° (c 1, CH₂Cl₂); IR (film): 1790 (C=O), 1750 cm⁻¹ (Ac); ¹H-NMR (CDCl₃): 2.19 (dd, 1H, J 3.4, 18.0 Hz, H-4), 2.38 (dd, 1H, J 9.1 Hz, H-4'), 3.77 (m, 1H, H-3), 3.78 (s, 3H, OCH₃), 3.81 (dd, 1H, J 6.3, 11.9 Hz, CH₄DAc), 3.88 (dd, 1H, J 3.9 Hz, CH₄DAc), 4.03, 4.08 (2d, 2H, -NCH₂-), 4.06 (m, 1H, -CHOTBDPS), 4.97 (dd, 1H, J 4.8, 5.9 HZ, B-CHOAC). $\frac{4}{4}$ (R¹=CH₂OAc, R²=A, R^{*}=H, R^{*}=Ac, R⁵=CH₂C₆H₄OCH₃): 80%; syrup; (α)_D -42.0° (c 0.6, CH₂Cl₂); IR (CHCl₃): 1790 (C=O), 1750 cm⁻¹ (Ac); ¹H-NMR (CDCl₃): 2.71 (dd, 1H, J 8.5, 17.5 Hz, H-4), 2.82 (dd, 1H, J 7.2 Hz, H-4'), 3.62 (ddd, 1H, J 3.1 Hz, H-3), 4.08, 4.19 (2d, 2H, -CH₂N), 4.13-4.21 (m, 2H, CH₂OAc, 5.18 (ddd, 1H, J 3.4, 5.0, 7.0 Hz, H=B), 5.24 (dd, 1H, H-\alpha). 4 (R¹=CH₂OAc, R²=R⁴=H, R³=OAc, R⁵=CH₂C₆H₄OCH₃): 90%; m.p. 127-130°C; (a) -76.4° (c 1, CH₂Cl₂); IR (nujol): 3380 (OH), 1800 (C=O), 1750, 1745 cm⁻¹ (Ac); ¹H-NMR (CDCl₃): 2.53 (dd, 1H, J 2.7, 17.9 Hz, H-2), 2.69 (dd, 1H, J 4.8 Hz, CH₄H₀OAc), 4.04 (m, 1H, =CHOH), 4.10, 4.21°(2d, 2H, =NCH₂-), 4.95 (dd, 1H, J 3.0, 4.5 Hz, =CHOAC). P. Deslongchamps, Stereoelectronic Effects in Organic Chemistry, Pergamon, Oxford, 1983, p. 221-241 <u>4</u> (R¹=R²=R³=H, R⁴=TBDPS, R⁵=CH₃): 78%; m.p. 70-72°C; IR (fi1m): 1795 cm⁻¹ (C=O); ¹H-NMR
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